

# Novel insertion of ethynylbenzene derivatives bearing an electron-withdrawing substituent at the 4-position into a P–C bond and their transannular insertion between a metal atom and an *ipso*-carbon atom of the phosphine ligand †

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Reactions of  $\text{Cp}^*\text{MCl}(\text{P}-\text{O})$  ( $\text{M} = \text{Rh}, \text{Ir}$ ) bearing P–O coordination with ethynylbenzene derivatives having an electron-withdrawing group in the presence of a  $\text{PF}_6^-$  anion result in either insertion into a P–C bond or a transannular addition between a metal and an *ipso*-carbon of the phosphine.

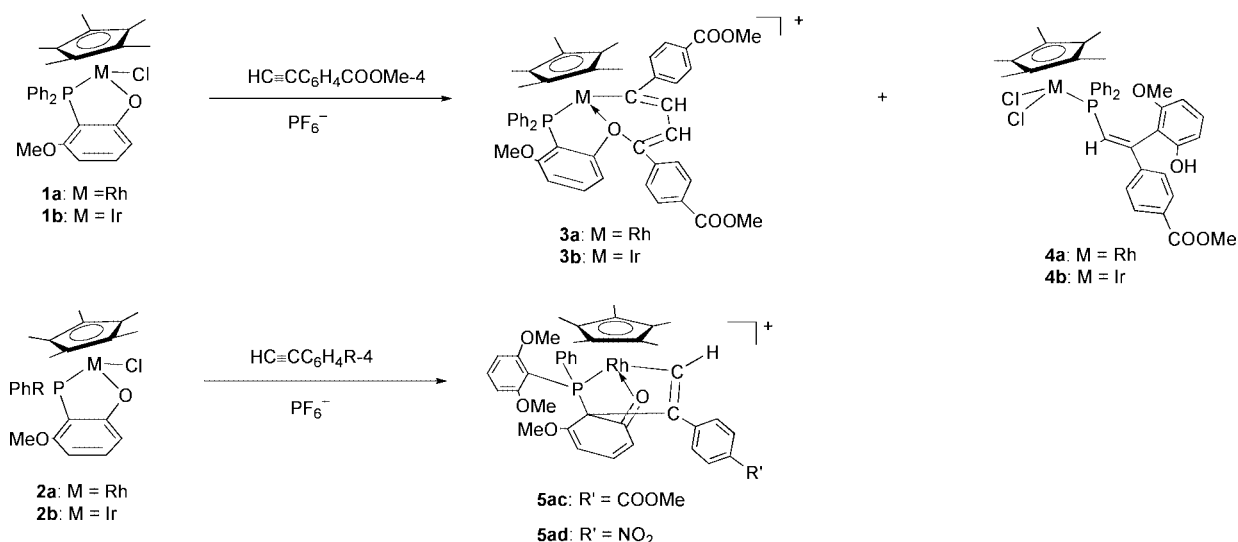
When transition metal halide complexes are allowed to react with alkynes in the presence of anions such as  $\text{PF}_6^-$ ,  $\text{BF}_4^-$ , etc., it is well-known that vinylidene or carbene complexes may be formed, depending on the solvent.<sup>1</sup> However, we have recently reported that the unprecedented insertion of alkyne into a metal–oxygen  $\sigma$ -bond can occur without the formation of vinylidene complexes.<sup>2</sup> On treatment of the rhodium complex  $[\text{Cp}^*\text{RhCl}(\text{MDMPP}-\kappa^2\text{P},\text{O})]$  **1a** ( $\text{MDMPP}-\kappa^2\text{P},\text{O} = \kappa\text{O}-1-\text{O}-\kappa\text{P}-2-\text{PPh}_2-3-\text{MeOC}_6\text{H}_3$ ) with mono- and di-substituted alkynes ( $\text{HC}\equiv\text{CPh}$ ,  $\text{HC}\equiv\text{CCOOR}$  or  $\text{ROCC}\equiv\text{CCOOR}$ ;  $\text{R} = \text{Me}, \text{Et}$ ) in the presence of the  $\text{PF}_6^-$  anion, single and double insertion of alkynes into the Rh–O bond occurred to form various types of complex, depending on the alkyne, as follows; (1) a complex bearing five- and six-membered rings arising from a double insertion, (2) a seven-membered metallacycle bearing a CO ligand arising from a single insertion and an extraction of CO from an ester and (3) a neutral complex bearing a seven-membered ring arising from a single insertion.<sup>2</sup>

When ethynylbenzene derivatives bearing an electron-withdrawing substituent at the 4-position of the phenyl group

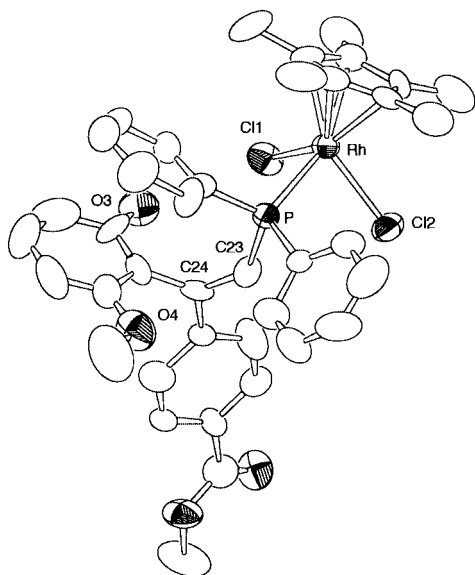
were treated with  $[\text{Cp}^*\text{MCl}(\text{MDMPP}-\kappa^2\text{P},\text{O})]$  (**1a**:  $\text{M} = \text{Rh}$ ; **1b**:  $\text{M} = \text{Ir}$ ) or  $[\text{Cp}^*\text{RhCl}(\text{BDMPP}-\kappa^2\text{P},\text{O})]$  **2a**<sup>5</sup> ( $\text{BDMPP}-\kappa^2\text{P},\text{O} = \kappa\text{O}-1-\text{O}-\kappa\text{P}-2-\text{PRPh}-3-\text{MeOC}_6\text{H}_3$ ) bearing a P,O bidentate ligand derived from (2,6-dimethoxyphenyl)diphenylphosphine (MDMPP) and bis(2,6-dimethoxyphenyl)phenylphosphine (BDMPP) in the presence of  $\text{KPF}_6$ , we found that novel types of reaction: an insertion into a P–C bond or a transannular addition between a Rh atom and an *ipso*-carbon atom of the phosphine ligand, occurred readily. We report here the reactions with ethynylbenzene derivatives bearing an electron-withdrawing substituent such as the COOMe or  $\text{NO}_2$  group.

Reaction of **1a** with  $\text{HC}\equiv\text{CC}_6\text{H}_4\text{COOMe}-4$  at room temperature in the presence of  $\text{KPF}_6$  in acetone– $\text{CH}_2\text{Cl}_2$  gave two compounds by crystallization from  $\text{CH}_2\text{Cl}_2$ –diethyl ether: **3a** ‡ and **4a** ‡ as confirmed from elemental analyses and FAB mass spectroscopy (Scheme 1). X-Ray analysis revealed that **3a** contained a (P,O,C) tridentate ligand resulting from the head-to-head double-insertion of alkyne into the Rh–O bond. § Two carbon atoms bearing the phenyl group were connected to the Rh and O atoms. The infrared spectrum showed bands due to methoxycarbonyl and  $\text{PF}_6^-$  groups at 1717 and 839  $\text{cm}^{-1}$ , respectively. The  $^1\text{H}$  NMR spectrum in  $\text{CD}_2\text{Cl}_2$  showed one doublet at  $\delta$  1.36 due to the  $\text{Cp}^*$  protons and three singlets at  $\delta$  3.35, 3.88 and 3.95; the former is due to the methoxy group and the others are due to the methoxycarbonyl groups. The  $^{31}\text{P}\{^1\text{H}\}$  NMR spectrum showed a doublet at  $\delta$  35.84 ( $J_{\text{RHP}} = 157.8$  Hz).

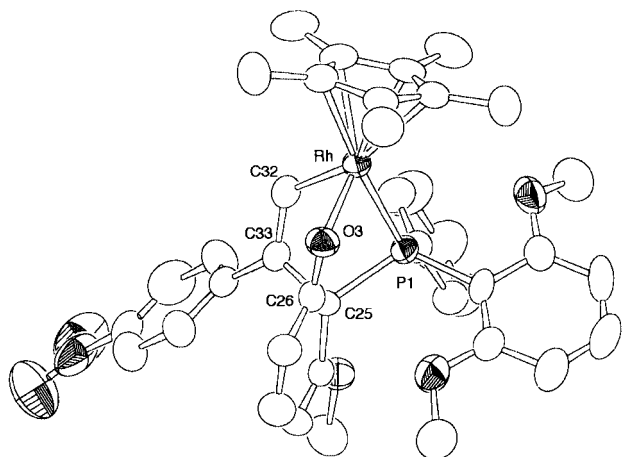
The infrared spectrum of **4a** showed three characteristic bands at 3285, 1717 and 1605  $\text{cm}^{-1}$  due to the OH, carbonyl groups and C–C double bond, respectively. In the  $^1\text{H}$  NMR spectrum a broad resonance due to a hydroxyl proton, which



Scheme 1 Reactions of **1** and **2** with  $\text{HC}\equiv\text{CC}_6\text{H}_4\text{R}'-4$  ( $\text{R} = 2,6-(\text{MeO})_2\text{C}_6\text{H}_3$ ,  $\text{R}' = \text{COOMe}$  or  $\text{NO}_2$ ).



**Fig. 1** Molecular structure of **3a**. Selected bond lengths (Å) and angles (°): Rh(1)–P(1) 2.341(5), Rh(1)–Cl(1) 2.428(5), Rh(1)–Cl(2) 2.411(5); P(1)–C(23) 1.83(2), C(23)–C(24) 1.32(2); P(1)–Rh(1)–Cl(1) 87.1(2), P(1)–Rh(1)–Cl(2) 85.0(2), Cl(1)–Rh(1)–Cl(2) 93.4(2), P(1)–C(23)–C(24) 135(1), C(23)–C(24)–C(25) 117(1).



**Fig. 2** Molecular structure of **5ad**. Selected bond lengths (Å) and angles (°): Rh(1)–P(1) 2.304(2), Rh(1)–O(3) 2.106(4), Rh(1)–C(32) 2.013(6), O(3)–C(26) 1.263(7), C(25)–C(26) 1.505(8), P(1)–C(25) 1.943(6), C(25)–C(33) 1.567(8), C(32)–C(33) 1.340(8); P(1)–Rh(1)–O(3) 81.1(1), P(1)–Rh(1)–C(32) 76.3(2), O(3)–Rh(1)–C(32) 85.2(2), Rh(1)–P(1)–C(25) 90.2(2), P(1)–C(25)–C(26) 105.1(4), C(25)–C(26)–O(3) 118.7(5), Rh(1)–O(3)–C(26) 117.9(4), Rh(1)–C(32)–C(33) 120.4(4), C(32)–C(33)–C(25) 114.2(5), P(1)–C(25)–C(33) 96.4(4).

disappeared on treatment with  $D_2O$ , appeared at  $\delta$  6.74 in addition to three characteristic methyl resonances at  $\delta$  1.41(d), 3.44(s) and 3.83(s), due to 1,2,3,4,5-pentamethylcyclopentadienyl, methoxy and methoxycarbonyl groups, respectively. There were no signs of a  $PF_6$  group in the infrared and  $^{31}P\{^1H\}$  NMR spectra, suggesting that the complex was neutral. The structure was confirmed by X-ray analysis (Fig. 1).<sup>¶</sup> As expected, the molecule is neutral and the rhodium atom is surrounded by two chlorine atoms and one phosphorus atom. *Cis*-insertion of alkyne into the P–C bond has occurred, accompanied by cleavage of a Rh–O bond. The Rh–P bond length of 2.341(5) Å is longer than that of **1a** by 0.04 Å, due to the absence of chelation and is 0.02 Å shorter than that (2.366(1) Å) of  $Cp^*RhCl_2(MDMPP)$ , probably due to lower steric demand than in the MDMPP ligand.<sup>3</sup> A similar reaction

occurred on treatment with **1b** in acetone– $CH_2Cl_2$  to give **3b** and **4b** (Scheme 1).<sup>‡</sup> The fact that a similar reaction in MeOH gave exclusively **3a** as the only isolated complex confirmed that the chlorine atom in **4** originated from dichloromethane.

Complex **2a** containing the bulky phosphine BDMPP was treated with  $HC\equiv CC_6H_4R-4$ , ( $R = COOMe$  or  $NO_2$ ) in acetone– $CH_2Cl_2$  in the presence of  $KPF_6$  at room temperature (Scheme 1) to yield complexes **5ac** (yellow,  $R = COOMe$ ) and **5ad** (brown,  $R = NO_2$ ) as confirmed from elemental analyses and FAB mass spectroscopy.<sup>‡</sup> X-Ray analysis of **5ad** revealed that the structure consists of five- and six-membered rings derived from a transannular insertion of alkyne between the Rh atom and the *ipso*-carbon atom of the phosphine ligand, accompanying the subsequent transformation of the Rh–O  $\sigma$ -bond to a Rh←O coordination (Fig. 2).<sup>||</sup> The change of this bonding mode caused an elongation of *ca.* 0.06 Å in the Rh–O bond length.

The IR spectrum showed bands due to methoxycarbonyl and ketone groups at 1715 and 1630  $cm^{-1}$  for **5ac** and due to a ketone group at 1628  $cm^{-1}$  for **5ad**, respectively. The  $^1H$  NMR spectrum showed three methoxy groups at  $\delta$  2.97 (s), 3.06 (bs) and 3.49 (bs) and one methoxycarbonyl group at  $\delta$  3.84 (s) for **5ac** and three methoxy groups at  $\delta$  2.99 (s), 3.08 (bs) and 3.52 (bs) for **5ad**, respectively. A remarkable feature is that the  $^{31}P\{^1H\}$  NMR doublets show large downfield shifts (*ca.*  $\delta$  140).

Further mechanistic studies are now in progress.

## Acknowledgements

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## Notes and references

<sup>‡</sup> Elemental analyses of all new complexes prepared here are in good agreement with the calculated values. Elementary analyses and spectroscopic data are enclosed in the electronic supplementary information.<sup>†</sup>  
<sup>†</sup> Crystal data for **3a**:  $C_{49}H_{47}O_6P_2F_6Rh$ ,  $M = 1010.8$ , monoclinic, space group  $P2_1/a$  (no. 14),  $a = 15.534(3)$  Å,  $b = 19.986(3)$  Å,  $c = 15.757(3)$  Å,  $\beta = 103.70(1)^\circ$ ,  $V = 4752(1)$  Å<sup>3</sup>,  $Z = 4$ ,  $D = 1.413$  g  $cm^{-3}$ ,  $\mu = 4.96$   $cm^{-1}$  (MoK $\alpha$ ),  $F(000) = 2072$ ,  $T = 298$  K. Data were collected on a Rigaku AFC5S diffractometer. The structure was solved by Patterson methods, and non-hydrogen atoms were refined anisotropically using full-matrix least-squares based on  $F^2$  to give  $R1 = 0.064$  for 3298 reflections and  $R_w = 0.198$  for 8328 reflections.

<sup>¶</sup> Crystal data for **4a**:  $C_{39}H_{42}O_6PCl_2Rh$ ,  $M = 811.5$ , orthorhombic, space group  $Pca2_1$  (no. 29),  $a = 34.835(7)$  Å,  $b = 9.135(8)$  Å,  $c = 12.305(5)$  Å,  $V = 3915(5)$  Å<sup>3</sup>,  $Z = 4$ ,  $D = 1.377$  g  $cm^{-3}$ ,  $\mu = 6.55$   $cm^{-1}$  (MoK $\alpha$ ),  $F(000) = 1672$ ,  $T = 300$  K. The structure was solved by Patterson methods, and non-hydrogen atoms were refined anisotropically using full-matrix least-squares based on  $F^2$  to give  $R1 = 0.068$  for 2216 reflections and  $R_w = 0.203$  for 3607 reflections.

<sup>||</sup> Crystal data for **5ad**:  $C_{39}H_{46}NO_6P_2F_6Rh$ ,  $M = 951.6$ , monoclinic, space group  $P2_1/a$  (no. 14),  $a = 15.299(7)$  Å,  $b = 14.826(7)$  Å,  $c = 19.505(9)$  Å,  $\beta = 106.78(3)^\circ$ ,  $V = 4235(3)$  Å<sup>3</sup>,  $Z = 4$ ,  $D = 1.492$  g  $cm^{-3}$ ,  $\mu = 5.57$   $cm^{-1}$  (MoK $\alpha$ ),  $F(000) = 1952$ ,  $T = 298$  K. The structure was solved by Patterson methods, and non-hydrogen atoms were refined anisotropically using full-matrix least-squares based on  $F^2$  to give  $R1 = 0.065$  for 4458 reflections and  $R_w = 0.180$  for 7708 reflections. CCDC reference number 186/2117. See <http://www.rsc.org/suppdata/dt/b0/b004984k/> for crystallographic files in .cif format.

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